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Chemical Process Principles: a Combined Volume Consisting of Part One Material and Energy Balances, Part Two Thermodynamics, Part Three Kinetics and Catalysis Hougen, Olaf A.; Watson, Kenneth M. Published by John Wiley & Sons, Inc. (1947)

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Principios de los procesos químicos (Hougen et al, tomo 1)

Most chemical processes are assumed to run at steady state, which implies that there are no variations of the process varies with time. Thus, rate of accumulation is identical zero under the condition of steady state. Thus, for a non-reacting species at steady state, the mass balance equation

Reading for Mass Balances - MIT

Hougen and Kenneth M. Watson were co-authors of the three-volume text titled Chemical Process Principles (Material and Energy Balances; Thermodynamics; Kinetics & Catalysis), which was a milestone in the literature of the profession because of its emphasis on applying the basic principles of

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mathematics, chemistry and physics to
the analysis of chemical processes.

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