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CHEMISTRY B- MOLES PACKET NAME: ____ HR: ____ PAGE 9 CHEMISTRY WORKSHEET # 6 MIXED MOLE PROBLEMS (GRAMS, MOLECULES, AND LITERS) You now know three things a mole can be: a molar mass, 6.02 x 10²³ molecules and, for a gas, 22.4 liters at STP.

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CHEMISTRY B- MOLES PACKET NAME: ____ HR: ____ PAGE 4 CHEMISTRY WORKSHEET # 2: THE MOLE AS A UNIT OF MASS Now that you know how to find the mass of one mole of a substance (molar mass) you can easily find the mass of several moles or the mass of a fraction of a mole using the factor-label technique.

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answer Molar Mass (# grams) Atomic Weight (for elements) Gram Formula/Molecular Weight (for compounds) 23 ATOMS (for elements) or 6.02 X 10²³ MOLECULES ... CHEMISTRY MOLES PACKET PAGE 8 — STP. = 2 = CHEMISTRY MOLES PACKET ...

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Chemistry B. Moles Packet INTRODUCTION TO MOLES. We are about to start on a unit of chemical calculations called "stoichiometry". Stoichiometry is how we calculate the relationships between the amounts of reactants and the amounts of products.

CHEMISTRY WORKSHEET # 2 MOLE PROBLEMS—THE MOLE AS A UNIT ...

CHEMISTRY B- MOLES PACKET NAME: ____ HR: ____ PAGE 3 CHEMISTRY WORKSHEET # 1 MOLAR MASS (GRAM MOLECULAR/FORMULA WEIGHTS) We know that grams are actually a measure of the mass of matter and not the weight. Mass is the quantity of matter present; weight is a measure of the pull of gravity on matter and is measured in pounds or newtons.

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A) 8.5 moles of sodium acetate dissolved in 1 L of water B) 5.5 moles of sodium acetate dissolved in 500 mL of water C) 1.8 moles of sodium acetate dissolved in 300 mL of water D) 1.2 moles of sodium acetate dissolved in 200 mL of water E) none of the above

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CHEMISTRY B- MOLES PACKET NAME: ____ HR: ____ PAGE 4CHEMISTRY WORKSHEET # 2: THE MOLE AS A UNIT OF MASSNow that you know how to find the mass of one mole of a substance (molar mass) you can easily find the mass ofseveral moles or the mass of a fraction of a mole using the factor-label technique.1 mol = a molar mass of an atom/molecule (g/mol)EXAMPLE: What is the mass of 5.00 moles of water H ...

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www.njctl.org Chemistry Mole Calculations 24. What is the molar mass of Iron (III) Nitrate? 25.)?What is the molar mass of Sucrose (C 12 H 22 O 11 26. How many grams are in 3 moles of Water? 27. How many grams are in 0.5 moles of Nitrogen Gas? 28. How many grams are in 12.3 moles of Dinitrogen Pentoxide? 29. How many grams are in 2.7 moles of ...

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