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Chemistry 11 Final Exam Study Guide Chemistry 11 - Final Exam Study Guide Page 18. An enthalpy diagram for an exothermicreaction: So, to summarize, in an exothermicreaction: ΔH is negative Products are lowerthan Reactants on the Enthalpy Diagram Heat is released to the surroundings. In an endothermicreaction: Heat is absorbedfrom the surroundings and converted into enthalpy.

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