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### **1.1 Chemistry in Context | General College Chemistry I**

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## **Chemistry in Context : Secondary: Oxford University Press**

Chemistry (4th Edition) Burdge, Julia  
Publisher McGraw-Hill Publishing  
Company ISBN 978-0-07802-152-7

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Chemistry in Context, 9th Edition by  
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chemistry in the atoms-first approach, CiC extracts chemistry from students' surroundings.

## **Chemistry in Context - Table of Contents - American ...**

Definitions for the book Chemistry in Context; Chapter 5; UCF CHM 1020.

Terms in this set (41) Electronegativity. is a measure of the attraction of an atom for an electron in a chemical bond. Polar Covalent Bond. a covalent bond in which the electrons are not equally shared but rather are closer to the more electronegative atom.

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(a) When two electrons are removed from the valence shell, the Ca radius loses the outermost energy level and reverts to the lower  $n = 3$  level, which is much smaller in radius. (b) The +2 charge on calcium pulls the oxygen much closer compared with K, thereby increasing the lattice energy relative to a less charged ion.

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