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Chapter 12 Stoichiometry 127 SECTION
12.1 THE ARITHMETIC OF EQUATIONS
(pages 353-358) This section explains
how to calculate the amount of
reactants required or product formed in
a nonchemical process. It teaches you
how to interpret chemical equations in

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terms of interacting moles,
representative particles, masses, and
gas volume at STP.

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05 CTR ch12 7/9/04 3:34 PM Page

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12.1 The Arithmetic of Equations > 11
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Chapter 12

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12.1 THE ARITHMETIC OF EQUATIONS (pages 353–358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

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12.2 Chemical Calculations > 13

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of things. • There are 6.022×10^{23}
atoms in 12.00 g ...

Chapter 3 Chemical Reactions and Reaction Stoichiometry

Chapter 3 Stoichiometry: Calculations
with Chemical Formulas and Equations
3.1 Multiple -Choice Questions 1) When
the following equation is balanced, the

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coefficients are 2, 25, 16, 18. $C_8H_{18} + O_2 \rightarrow CO_2 + H_2O$
A) 2, 3, 4, 4 B) 1, 4, 8, 9 C) 2, 12, 8, 9 D) 4, 4, 32, 36 E) 2, 25, 16, 18
Answer: E Diff: 2 Page Ref: Sec. 3.1

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12.3 Limiting Reagent and Percent Yield
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